MP100274 A-US

## Declaration under 37 CFR 1.132 with regard to US utility patent application number 10/675,138

- I, Alan Reginald Minihan, a British subject of 18 Green Lane, Wallasey, Merseyside, U.K. declare the following:
- 1. I hold the degree of D. Phil. in Chemistry from University of Oxford and the degree of Master of Arts from University of Oxford. I am a Chartered Chemist and a member of the Royal Society of Chemistry.
- 2. I am presently employed as Group Product Development Manager by Ineos Silicas Limited, Bank Quay, Warrington, UK and have worked for a total of 21 years on the chemistry and structure of inorganic chemicals for Unilever plc, Crosfield Ltd, and Ineos Silicas Ltd.
- 3. The research work detailed below was carried out as part of a joint BRITE (EU-sponsored) research project (no F14W-CT95-0016) between British Nuclear Fuels Ltd. (UK), Crosfield Ltd. (UK), IVO International (Finland), University of Helsinki (Finland) University of Salford (UK) between 1st January 1996 and 31st December 1998. The table of data, Table 26, annexed to this document is from the final report detailing the work carried out in the project.
- Crosfield ltd. changed its name to Incos Silicas ltd. on 13th March 2001.
- 5. Table 26, which is annexed to this document, shows the distribution coefficients (Kd) for various isotopes in acid solution for tungsten (W)-doped antimony silicates (WSS samples HMS18) in comparison to antimony silicate (HMS10) and titanium (Ti) doped antimony silicate (HMS19). Description of the preparation of the materials tested is also included in table 26.
- 6. From the comparative data presented in the table it can be seen that the tungsten doped antimony silicates give much higher values for Kd (e.g 702, 8182, 14251-18303 for HMS18a1; i.e. good extraction behaviour) in comparison to the Kd values obtained for the titanium-doped antimony silicate (22, 187, 1.06). It is believed that the test results indicated herein are representative of the testing program, even though there may be other tests, not included herein, that may have been conducted in the time frame of the program.
- 7. From these data it was concluded that Ti was an undesirable dopant for antimony silicate to be used for extraction of radioactive metals from acid solution and work on this dopant was not progressed.
- 8. I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

09 May 2006

MP100274 A-US

Signature

12486455418

Date 09 May 2006

Name Alan Reginald Minihan

12486455418

Table 26. Distribution coefficients (Kd) for W doped antimony silicates

Ç	,	Î	,	4	w		•		1	٥	•			L.		2	_	}
HMS18c1	DEBOTCEATU	HMS18a3	HMS18a2	HMS18a1d	HMS18a1	WSS	HMS17c3	THE TANK	UKIONA	HMS17c1	HMS17e2		HMS17a1	Moss	TiSS	(SpSi)	(KSS)	Sample name
+Na, WO, *2H,O	+Na,WO,*2H,O	+Nb,WO,72H,O	+Na, WO, *2H,O	HMS12	HWS12 +Na <sub>2</sub> WO <sub>4</sub> *2H <sub>2</sub> O		+ NH <sub>4</sub> (MoO <sub>3</sub> ) <sub>2</sub>	+ NH4(MaO2)2	+ NH,(MoO,),	OI SZWIH	+ NH <sub>4</sub> (MoO <sub>3</sub> ) <sub>2</sub>	+ NH4(M0O3)2	HMS12	HMS10+11CA		SbCl <sub>5</sub> in 4 M HCl <sub>6</sub> Na <sub>2</sub> Si <sub>3</sub> O <sub>7</sub> (Fluka)	KSb(OH), TEOS,	Starting materials
1:1:1 (weigh.)	J.I.v.I (weign.)	1:1:2 (weigh.)	1:1:1 (weigh.)	1:1:0.5 (weigh.)	1:1:0.5 (weigh.)		1:2.5:0.5 (sol)	1.2.3.1.7 (801)	1.2 61 9 61	1:1:1:1 (sol)	1:1:0.2 (weigh)		[:]:] (weigh.)	1:1:0:51 (501)		solutions)	i:I (weighed)	Starting Sb:Si:W ratio
20h at 77C	let to dry at I/C		at 77C overnight	let to dry at 77C	1% mixture at 77C			a //C, Overlight		at 77C, 2 days	i		at 77C, 3 days	at puc, ta		1% mixture at 77C overnight	1% mixture at 77C, overnight	Preparation method
Semicryst. SbSi	Amorphous	Cryst. Onknown	Amorphous	Amorphous	Amorphous		Amorphous	Amorpaous		Amorphous	Amorphous		Crystalline, AMP9	Amorphous		Crystalline (as antimonic acid)	Аторнаш	XRD trace
10652	1332	85.8 (dissolves)	272	670	702		205	220		811	472		400	72		1354-3702	590	in 0.1 M HNO,
3252	4[382	282 (dissolves)	2489	8168	8182		811	199		141	2012	Č	122	187		35515-102700	19660	<sup>83</sup> Sr Kd ml/gj in 0.1 M HNO,
859	1762	48.5 (dissolves)	251	1320	14251-18303		60	116	į	75	2012	8	300	1.06		1509-4282	1959	o 0.1 M HNO3

-	5	. ☆	14	£1.	12	<u>, , , , , , , , , , , , , , , , , , , </u>	01	9
COS 1 CTAILS	HMS18c7d	HW818c6*	HMS18c5	HMS18c4	HMS18c3d	HMS18c3	HMS18c2d	HMS18c2
+Na <sub>2</sub> WO <sub>4</sub> *2H <sub>2</sub> O	HMS10 +NB,WO,*2H,O	+Na₂WO,*2H₂O	+NB,WO,*2H,O	+Na <sub>2</sub> WO,*2H <sub>2</sub> O	HMSIO +Na <sub>2</sub> WO <sub>4</sub> *2H <sub>2</sub> O	HMS10 +Na <sub>2</sub> WO <sub>4</sub> *2H <sub>2</sub> O	+Nb <sub>2</sub> WO <sub>4</sub> *2H <sub>2</sub> O	HMS10 +Na <sub>2</sub> WO <sub>4</sub> *2H <sub>2</sub> O
1:2.5:1.7 (301)	1:2.5:1.7 (sol)	(	1:2:5:0:5 (sol)	1.2.5:1 (sol)	0.5:3:1 (weigh.)	0.5:1:1 (weigh.)	1:1:1 (weigh.)	1:1:1 (weigh.)
I d at 77C	led to dry at 77C	_n_	<b>:</b>	l d at 77C	let to dry at 77C	2 days at 77C	let to dry at 77C	2 days at 77C
Cryst. SbSi (int?).	Cryst. SbSi	Cryst S6SI	Cryst. S5Si	Cryst. SbSi	Cryst, SbSi	Cryst. SbSi	Cryst. SbSi	Cryst SbSi
3142	6924	18697	7008	15356	20970	17745	17188	1444)
435	5221	42151	9492	9651	8075	5637	35499	4608
76.4	289	1440	1011	611	640	224	1515	1552

137